

Bio-organic Chemistry, CHEM 300  
Spring 2008

Lecture: 1:00 p.m., MWF, MAC 105 (class #3427)

Prerequisite: C or better in CHEM 212 or CHEM 108

Instructor: Dr. W. Johnson  
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office hours: 11:00-11:50 a.m., MWF  
or by appointment

Web site: <http://www.uwgb.edu/johnsonw/>

Course materials:

**Fundamentals of Organic Chemistry**, 6th ed., McMurry and Simanek, Thomson, Brooks/Cole, 2007.

**Study Guide and Solutions Manual for McMurry and Simanek's Fundamentals of Organic Chemistry**, 6th ed., McMurry, Thomson, Brooks/Cole, 2007.

**The Molymod™ Molecular Model Set for Organic Stereochemistry**, Indigo® Instruments. **Online Web-based Learning (OWL)**, purchase with new book or separately.

To purchase OWL separately (\$28.49) go to the Phoenix Bookstore web page:

<http://www.thephoenixbookstore.com> and click on **ichapters.com** (found in right column, near bottom of page). In the **Basic Search** box enter this ISBN: **0495051012**. Click on **My Cart** in upper right corner and then checkout.

To register OWL after you have purchased an access code go to:

<http://owl.thomsonlearning.com/> At the "Login or Register" choose **Organic** and then select our textbook (see above title). Choose **University of Wisconsin – Green Bay** and then choose the **Student Registration** link. After registering you are ready to use OWL.

To use OWL after you have registered go to: <http://owl.cengage.com/owl-c/user/loginpage.cgi?Server=owl-univwisconsin-greenbayfundo&UserType=Student> and enter your login and password.

The purpose of this course is for you to learn the fundamental chemistry of the molecules of life in one semester. These organic molecules contain a variety of different functional groups. After completion of this course you ought to understand the molecular structure and the reactivity of organic molecules with precision and biology ought to become more understandable at the molecular level.

Working out the problems in the text is highly recommended! In addition to problems in the text, problems at OWL are assigned for each chapter. There are 150 points for problems assigned in OWL (10 points for each chapter). There will be 4 one-hour exams (400 points), a two page article on the organic chemistry of a topic of interest (100 points), and a final exam (200 points), which will contribute to your grade for the course (850 points total). Make-up examinations will be allowed **ONLY** in the case of illness or serious tragedy and permission must be obtained for a make-up exam prior to the scheduled examination. The expected range of percentage scores for grades in this course is: A, 91-100; AB, 86-90; B, 75-85; BC, 68-74; C, 55-67; D, 45-54; F, 0-44. This grade scale may be adjusted at any time by the instructor if a different scale is deemed more appropriate for the nature of the course during this particular semester.

This is a demanding course and will require disciplined serious study in order to master the material and obtain an acceptable grade. A minimum of 10 hours per week of study outside of the classroom is needed.

**Two Page Article:** You will prepare a two-page article on the organic chemistry of a biological process or other topic of interest. Select a topic and clear it with me for approval by Monday, March 10. Only one person in the course may present a particular topic. A list of some possible topics is found on the next page.

Each article is to have the following parts: title, author, main body, and references. Design your article to effectively teach the organic chemistry of your topic to the other students in this course. You may wish to use columns, different fonts, and images to enhance the appearance of your article. Your article is due before midnight on Monday, April 28. Your article needs to be formatted either as a Word document with a “.doc” extension or as a portable document file (.pdf). You then need to submit it at the course D2L site at one of the three discussion topics (nutrition, medicine, and other) at the "Discussions" page.

You may earn up to 15 bonus points for reading and evaluating the articles prepared by other students and then submitting your nominations for 1) most interesting, 2) best designed, and 3) best illustration of chemistry in each of the three topic groups (nutrition, medicine, and other). You will then have a total of nine nominations for an award, three nominated papers in each of three topic groups. All student articles will be available for you to read and evaluate at the D2L site. You will be able to submit your nominations for award winning articles starting on Tuesday, April 29 at the D2L "Discussions" site. **All nominations must be made before midnight, Wednesday, May 7.**

Potential Topics for your Article:

acetaminophen mechanism of action	isotope effects
acrylamide in baked foods	Kevlar, nylon, and other polyamides
alcohol dehydrogenase	lysozyme
aldolase	methylation of DNA by dimethylsulfate (DMS)
aminoacyl-tRNA synthetases	nitrosamines in food and health concerns
ampicillin	nonenzymatic peptide synthesis
anabolic steroids	oxidative damage of DNA (mutation mechanisms)
antidotes for poisonous war gas	particular metabolic reactions
aspirin mechanism of action	penicillin
base catalyzed RNA cleavage	pepsin
breath analyzer chemistry	phosphocreatine
cancer chemotherapy	polycarbonates
chiral drugs	polyesters
cross-linking reactions used in biology	polyurethanes
cytochrome P-450 chemistry	preparation of affinity resins for chromatography
dioxin	puromycin
DEAT and how it works	ribonuclease A
DNA methylases	ribonuclease P
DNA modification reactions	ritalin mechanism of action
enzymes with interesting mechanisms	role of a particular vitamin
epoxy resins	self-splicing introns
ethylnitrosourea reaction with DNA	spliceosome
food preservatives	sun screen and how it works
gene machine chemistry	tetracycline
homocysteine	thermosetting polymers
how a particular drug is synthesized	trans fats
how a particular drug works	triosephosphate isomerase
hydrazine reaction with DNA	trypsin
hydrazine reaction with protein	valium mechanism of action
insecticides	vitamin A, B1, etc., choose one
iodination of proteins for radio-labeling	

### Tentative Lecture Schedule

	Date		Lecture	Chapter	Topic
Jan	23	W	1	1	Lewis structures, hybrid orbitals
	25	F	2	1	acids and bases
	28	M	3	2	functional groups, alkanes
	30	W	4	2, 16.5	cycloalkanes, steroids
Feb	1	F	5	3	alkenes
	4	M	6	3	addition of HCl, isoprene and terpenes
	6	W	7	3	
	8	F	8	4	reactions of alkenes
	11	M			<b>Exam #1</b> (Lectures 1 → 8)
	13	W	9	4	reactions of alkenes
	15	F	10	4	
	18	M	11	4	reactions of alkynes
	20	W	12	5	aromatic compounds
	22	F	13	5	
	25	M	14	5	substituent effects
	27	W	15	5	
	29	F	16	6	stereochemistry
Mar	3	M			<b>Exam #2</b> (Lectures 9 → 16)
	5	W	17	7	alkyl halides
	7	F	18	7	S <sub>N</sub> 1 and S <sub>N</sub> 2 reactions
	10	M	19	7	E1 and E2 reactions
	12	W	20	8	alcohols, phenols, and ethers
	14	F	21	8	
					Spring Recess
	24	M	22	8	
	26	W	23	8	phenols, thiols, and sulfides
	28	F	24	9	aldehydes and ketones (names, synthesis, oxidation, and reduction; through 9.4)
	31	M			<b>Exam #3</b> (Lectures 17 → 24)
Apr	2	W	25	9	nucleophilic addition reactions
	4	F	26	10	carboxylic acids

Date			Lecture	Chapter	Topic
Apr	7	M	27	10	carboxylic acids
	9	W	28	10	derivatives of carboxylic acids
	11	F	29	10	carbonyl $\alpha$ -H reactivity
	14	M	30	11	$\alpha$ -substitution reactions
	16	W	31	11	
	18	F	32	12	amines
	21	M			<b>Exam #4</b> (Lectures 25 $\rightarrow$ 32)
	23	W	33	13	structure determination
	25	F	34	13	
	28	M	35	13	
	30	W	36	14	carbohydrates
May	2	F	37	14	
	5	M	38	14	
	7	W	39	16	lipids
	9	F	40	16	
	12	M	41		summary
	19	M			Final Exam, 1:00 p.m.